

Total Synthesis of the Tetraquinane Diterpenoid (\pm)-Crinipellin B

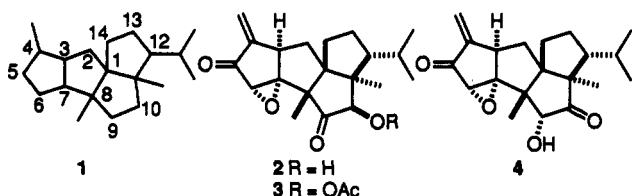
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Received October 14, 1992

Summary: Beginning with 2-methyl-2-cyclopenten-1-one (5), the total synthesis of the structurally novel tetraquinane diterpenoid (\pm)-crinipellin B (4) has been accomplished via a 22-step sequence of reactions.

The crinipellane-type diterpenoids are a small family of structurally unprecedented natural products that share the 12-isopropyl-4,8,11-trimethyltetracyclo[6.6.0.0^{1,11}.0^{3,7}]-tetradecane skeleton 1.¹ Three members of this group, crinipellin A, *O*-acetylcrinipellin A, and crinipellin B, isolated from different strains of the fungus *Crinipellis stipitaria* (Agaricales), have been shown to possess the structures shown in formulas 2-4, respectively.¹ Each of these highly oxygenated substances contains eight stereogenic centers, three of which consist of contiguous quaternary chiral carbon atoms (C-1, C-8, C-11). Compounds 2-4 exhibit interesting biological properties, including powerful antibiotic activity.^{1,2}



A number of reports describing synthetic approaches to the crinipellins have appeared.³ However, to our knowledge, the total synthesis of one or more of these substances has not yet been recorded. We describe herein a total synthesis of (\pm)-crinipellin B via a route in which two new 5-membered ring annulation methods play key roles.

Conversion of the enone 5 into the functionalized tetraquinane 18 was achieved via the reaction sequence summarized in Scheme I. Copper(I)-catalyzed conjugate addition of *i*-PrMgBr to 5 in the presence of Me₃SiCl and HMPA provided the enol silyl ether 6.⁴ As expected (steric approach control), alkylation of the lithium enolate derived from 6 with 2-(bromomethyl)-1-butene in the presence of (Ph₃P)₄Pd⁵ gave, stereoselectively, the monoalkylated product 7, along with minor amounts of dialkylated materials. Base-promoted cyclization of the dione 8, which was readily obtained by oxidative cleavage of the alkene function in 7,⁶ produced the bicyclic enone 9 in excellent yield.

Transformation of 9 into the triquinane 12 made use of a new annulation sequence developed recently in our

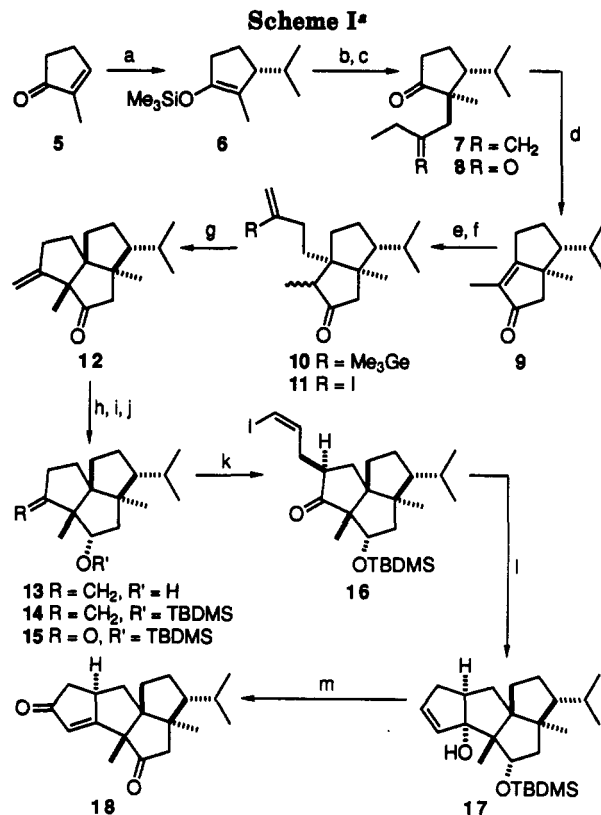
(1) Anke, T.; Heim, J.; Knoch, F.; Mocek, U.; Steffan, B.; Steglich, W. *Angew. Chem., Int. Ed. Engl.* 1985, 24, 709.

(2) Kupka, J.; Anke, T.; Oberwinkler, F.; Schramm, G.; Steglich, W. *J. Antibiot.* 1979, 32, 130.

(3) Mehta, G.; Rao, K. S. *J. Chem. Soc., Chem. Commun.* 1987, 1578. Mehta, G.; Rao, K. S.; Reddy, M. S. *Tetrahedron Lett.* 1988, 29, 5025. Schwartz, C. E.; Curran, D. P. *J. Am. Chem. Soc.* 1990, 112, 9272. Mehta, G.; Rao, K. S.; Reddy, M. S. *J. Chem. Soc., Perkin Trans. 1*, 1991, 693.

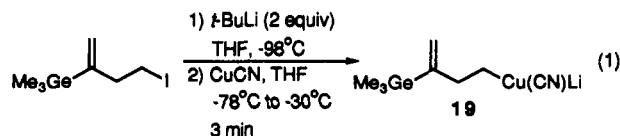
(4) All new compounds were spectroscopically characterized and gave satisfactory elemental analyses and/or high-resolution mass spectrometric molecular mass determinations.

(5) Negishi, E.; John, R. A. *J. Org. Chem.* 1983, 48, 4098.



(a) *i*-PrMgBr, CuBr·Me₂S, Me₃SiCl, HMPA, THF, -78 °C, 4 h; Et₃N (94%); (b) MeLi, THF, 0 °C; 2-(bromomethyl)-1-butene, (Ph₃P)₄Pd, THF, -20 °C, 2 h; 0 °C, 5 h (76%); (c) O₃, MeOH-CH₂Cl₂, -78 °C; Me₂S, -78 °C to room temperature; concentrate mixture, add 10% HCl-H₂O and THF, stir at room temperature for 18 h (93%); (d) MeONa, MeOH, reflux 15 h (97%); (e) reagent 19, Me₃SiBr, THF, -78 °C, 8 h; -48 °C, 2 h (83%); (f) I₂, CH₂Cl₂, room temperature, 16 h (98%); (g) (Ph₃P)₄Pd (21 mol %), *t*-BuOK, *t*-BuOH, THF, room temperature (84%); (h) *n*-Bu(*i*-Bu)₂Al(H)Li, Et₂O, -78 °C, 2 h; 0 °C, 1 h (93%); (i) *t*-BuMe₂SiOSO₂CF₃, Et₃N, CH₂Cl₂, -78 °C, 75 min; 0 °C, 20 min (98%); (j) OsO₄, C₅H₅N, room temperature, 23 h; NaHSO₃, H₂O, 1 h; Pb(OAc)₄, THF, 0 °C, 30 min; HOCH₂CH₂OH, 10 min (93%); (k) *i*-Pr₂NLi, THF, -78 °C; (*Z*)-3-bromo-1-iodopropene, room temperature, 7.5 h (76%); (l) *n*-BuLi (2.5 equiv), THF, -78 °C, 110 min (93%); (m) C₅H₅N·CrO₃·HCl, CH₂Cl₂, Celite, room temperature, 3.5 h (51%).

laboratory.⁷ Thus, conjugate addition of the novel cuprate reagent 19 (see eq 1) to the enone 9 in the presence of



Me₃SiBr, followed by appropriate workup, afforded the keto trimethylgermane 10 (mixture of epimers), which was

(6) The ¹H NMR spectrum of the crude product acquired from subsection of 7 to ozonolysis-Me₂S (step c, Scheme I) showed MeO signals, indicating the presence of a ketal function. Acid hydrolysis of this material produced 8 in high yield.

(7) Piers, E.; Marais, P. C. *J. Org. Chem.* 1990, 55, 3454.

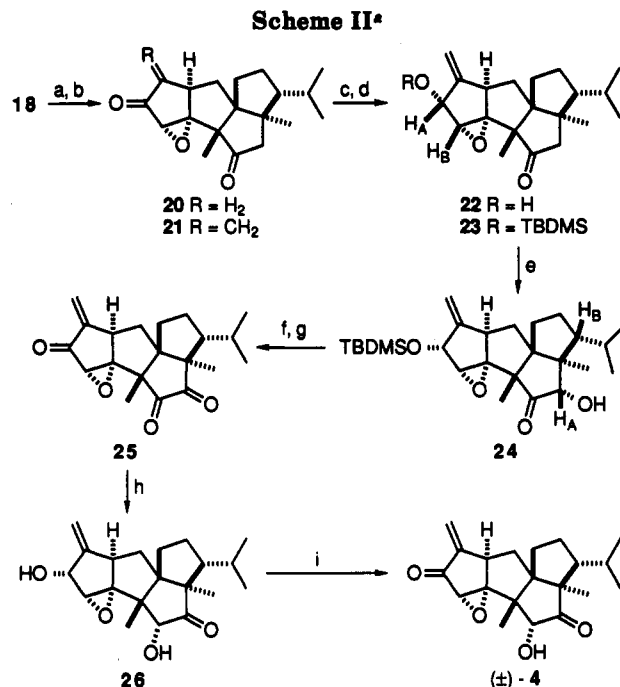
converted smoothly into the corresponding keto iodide 11. An efficient Pd(0)-catalyzed cyclization of 11 in the presence of *t*-BuOK⁷ gave 12. The overall conversion of 5 into 12 was highly stereoselective and produced a functionalized tricycle in which the three contiguous quaternary stereogenic centers required for the eventual synthesis of (±)-crinipellin B (4) had been installed cleanly and efficiently.

A straightforward sequence of reactions served to effect the efficient conversion of 12 into the keto ether 15 (mp 47–48.5 °C), via the intermediates 13 and 14. Of particular note was the highly stereoselective reduction of 12 with the complex metal hydride derived from reaction of *i*-Bu₂AlH with *n*-BuLi.

Construction of the fourth required 5-membered ring was initially problematic. Although a number of known cyclopentenone annulation methods⁸ were attempted, none were successful. In the present case, the usual difficulties associated with the conversion of (unsymmetrically substituted) cyclopentanones into the corresponding bicyclo[3.3.0]oct-1-en-3-ones are exacerbated by the hindered nature of the carbonyl function in 15. In view of these difficulties, a new cyclopentenone annulation sequence was developed (transformation of 15 into 18, Scheme I).⁹

Alkylation of 15 with (*Z*)-3-bromo-1-iodopropene¹⁰ was highly stereoselective and provided the keto iodide 16 efficiently. Treatment of 16 with *n*-BuLi in THF at –78 °C¹¹ produced, in excellent yield, the allylic alcohol 17 (mp 61.5–63 °C). Completion of the necessary annulation sequence required oxidative rearrangement of 17 with a Cr(VI) reagent. Interestingly, treatment of 17 with excess PCC in the presence of Celite¹² produced, as the major product,¹³ the keto enone 18 (mp 130–131.5 °C). Thus, in addition to effecting the expected process (cyclopentenone formation), this reaction also caused oxidative conversion of the silyl ether function into the corresponding carbonyl group.

Completion of the synthesis of (±)-4 was carried out as summarized in Scheme II. The epoxide 20, readily derived from 18, was treated sequentially with (Me₃Si)₂NLi and dimethyl(methylene)ammonium iodide.¹⁴ Flash chromatography of the resultant amino ketone effected elimination of Me₂NH and produced the enone 21¹⁵ (mp 157.5–158.5 °C). Thus, the stereoselective construction of the



^a (a) H₂O₂, NaHCO₃, H₂O–THF (1:2), room temperature, 55 min (84%); (b) (Me₃Si)₂NLi, THF, –78 °C; (H₂C=NMe₂)⁺I[–], –78 °C, 70 min; –70 °C, 18 min; flash chromatography (silica gel) (78%); (c) NaBH₄, MeOH–THF, –78 °C, 85 min; –63 °C, 15 min (80%); (d) *t*-BuMe₂SiOSO₂CF₃, Et₃N, CH₂Cl₂, –78 °C, 2 h; 0 °C, 70 min (88%); (e) (Me₃Si)₂NK, THF, –78 °C; 2-(phenylsulfonyl)-3-phenyloxaziridine, –78 °C, 45 min (68%); (f) *n*-Bu₂AlF, THF, room temperature, 75 min; (g) Dess–Martin periodinane reagent (ref 18) (4 equiv), C₆H₅N (2 equiv), CH₂Cl₂, room temperature; Na₂S₂O₃, NaHCO₃, H₂O (44% from 24); (h) *n*-Bu(*i*-Bu)₂Al(H)Li (4.6 equiv), Et₂O–THF, –78 °C, 30 min (41%); (i) C₅H₅N–SO₃, Me₂SO, Et₃N, CH₂Cl₂, room temperature, 9.5 h (49%).

complete carbon skeleton of (±)-crinipellin B (4) had been achieved. Intermediate 21 was smoothly transformed, via the alcohol 22¹⁶ (mp 146–147.5 °C), into the silyl ether 23 (mp 187–188.5 °C). The last required oxygen function was introduced by hydroxylation¹⁷ of the potassium enolate of 23, a process that afforded, stereoselectively, the α-hydroxy ketone 24¹⁶ (mp 190.5–192 °C). It may be noted that 24 possesses the relative configuration at C-10 opposite to that present in crinipellin A (2).

Cleavage of the silyl ether function in 24, followed by immediate oxidation¹⁸ of the resultant product, gave the triketone 25 (yellow-orange needles, mp 188–189.5 °C). Attempts to effect chemoselective reduction of only the C-9 carbonyl group in 25 were not successful. However, low-temperature reduction of this material with *n*-Bu(*i*-Bu)₂Al(H)Li provided the keto diol 26¹⁹ as the major product, along with less polar (TLC) byproducts. Oxidation²⁰ of 26 afforded, in addition to some starting material (12%) and a minor byproduct, (±)-crinipellin B (4). This substance (mp 153.5–155 °C) exhibited spectra in full accord with structural formula 4, and its ¹H NMR

(8) Paquette, L. A. *Top. Curr. Chem.* 1979, 79, 41; 1984, 119, 1. Paquette, L. A.; Doherty, A. M. *Polyquinane Chemistry (Reactivity and Structure Concepts in Organic Chemistry)*; Springer-Verlag: Berlin, 1987; Vol. 26. Hudlicky, T.; Price, J. D. *Chem. Rev.* 1989, 89, 1467.

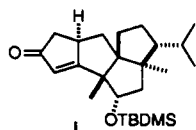
(9) The generality of this new, potentially useful method is currently under investigation.

(10) This substance was prepared by reduction (*i*-Bu₂AlH, THF) of methyl (*Z*)-3-iodopropenoate (Ma, S.; Lu, X.; Li, Z. *J. Org. Chem.* 1992, 57, 709), followed by reaction of the resultant alcohol with Ph₃PBr₂ in CH₂Cl₂.

(11) Cf. Piers, E.; Marais, P. C. *Tetrahedron Lett.* 1988, 29, 4053.

(12) Paquette, L. A.; Leone-Bay, A. *J. Am. Chem. Soc.* 1983, 105, 7352.

(13) Minor products from this oxidation were **1** and the diketone epoxide **20** (see Scheme II). Interestingly, **1** did not undergo epoxidation when treated with H₂O₂ in the presence of base (Scheme II, step a).



(14) Schreiber, J.; Maag, H.; Hashimoto, N.; Eschenmoser, A. *Angew. Chem., Int. Ed. Engl.* 1971, 10, 330.

(15) The structure of 21, and thereby of previous intermediates in the synthetic sequence, was confirmed by an X-ray crystallographic study. We are grateful to Dr. S. J. Rettig for carrying out this structure determination. Details will be reported elsewhere.

(16) The relative configuration of the newly introduced stereogenic center of this substance was confirmed by ¹H NMR spectroscopy. Thus, in nuclear Overhauser enhancement difference experiments, irradiation of H_A (δ 4.52 for 22, 4.04 for 24) caused enhancement of the signals due to H_B (δ 3.52 for 22, 1.27–1.37 (m) for 24).

(17) Davis, F. A.; Vishwakarma, L. C.; Billmers, J. M.; Finn, J. *J. Org. Chem.* 1984, 49, 3241.

(18) Dess, D. B.; Martin, J. C. *J. Org. Chem.* 1983, 48, 4155.

spectrum was found to be identical with that of natural (-)-crinipellin B.²¹

In summary, a 22-step conversion of 2-methyl-2-cyclopenten-1-one (5) into (\pm)-crinipellin B (4) was achieved.

(19) The overall conversion of 24 into 26 could also be achieved as follows: (i) oxidation of 24 with tetra-*n*-propylammonium perruthenate and *N*-methylmorpholine *N*-oxide in CH₂Cl₂ (Griffith, W. P.; Ley, S. V. *Aldrichim. Acta* 1990, 23, 13) (74%); (ii) silyl ether cleavage with *n*-Bu₄NF in THF (68%); (iii) reduction of the C-9 carbonyl group with *n*-Bu-(*i*-Bu)₂Al(H)Li in Et₂O-THF (53%). Although this sequence was somewhat more efficient than that summarized in Scheme II, the intermediates were quite unstable and, therefore, were difficult to purify and completely characterize.

(20) Parikh, J. R.; Doering, W. von E. *J. Am. Chem. Soc.* 1967, 89, 5505.

(21) We are very grateful to Professor W. Steglich for sending us a copy of the ¹H NMR spectrum of natural (-)-crinipellin B.

Two new annulation methods developed in our laboratory played important roles in the assembly of the required tetraquinane carbon skeleton (see conversions 9 into 12 and 15 into 18).

Acknowledgment. We are grateful to NSERC of Canada for financial support and for a Postgraduate Scholarship (to J.R.). We also thank FCAR, Quebec, and Bio-Mega Inc., Laval, Quebec, for Scholarships (to J.R.).

Supplementary Material Available: Experimental procedures for the preparation and ¹H NMR spectra (400 MHz, CDCl₃) of compounds 6-18, 20-26, and (\pm)-4, experimental details of the X-ray crystallographic study on intermediate 21, and a stereoview of this substance (40 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.